

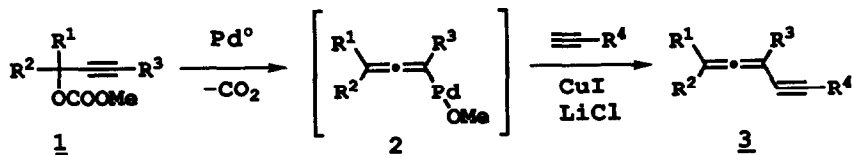
PALLADIUM-CATALYZED REACTIONS OF 2-ALKYNYL CARBONATES WITH TERMINAL ACETYLENES: A NEW SYNTHETIC METHOD FOR 1,2-DIEN-4-YNES

Tadakatsu MANDAI, Tatsuya NAKATA, Hiroshi MURAYAMA, Hiromasa YAMAOKI, Masakazu OGAWA, Mikio KAWADA, and Jiro TSUJI*

Department of Applied Chemistry, Okayama University of Science
Ridai-cho, Okayama, 700, Japan

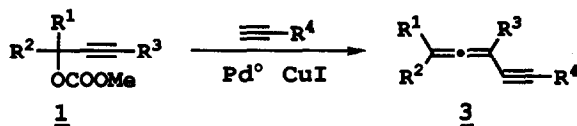
Summary; The reaction of substituted 2-alkynyl carbonates with terminal acetylenes in the presence of palladium-phosphine complex and copper(I) iodide proceeds to give 1,2-dien-4-yne.


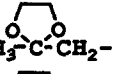
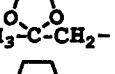
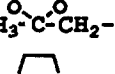
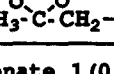
In our continuing studies on the palladium catalyzed transformations of 2-alkynyl carbonates **1**, which proceed by the formation of allenyl palladium complex **2** as an intermediate,¹ we have attempted their reaction with terminal acetylenes, expecting the formation of 1,2-dien-4-yne **3** (allenyl acetylenes). Allenyl acetylenes are present in naturally occurring compounds produced by some microorganisms,² and their simple synthetic method is highly desirable. Also this study is a part of our synthetic studies toward cumulative ene-yne-allene systems which are attracting attention for their DNA cleavage action.³ The known synthetic method is the palladium-catalyzed reaction of allenyl bromides with acetylenes or zinc acetylides.⁴ As expected, we observed the formation of 1,2-dien-4-yne **3** as shown below, and the results are presented in this paper.



The reaction of alkenyl or aryl halides with terminal acetylenes catalyzed by palladium complex and CuI is known.⁵ Thus we attempted reaction of 2-alkynyl carbonates with terminal acetylenes in the presence of a palladium phosphine complex and CuI without success. Then we tried the reaction in the presence of an excess of LiCl and observed a successful coupling reaction with acetylenes. A typical example is shown below. (Entry 9 in the Table) In THF (3 ml) were added CuI (20.2 mg, 0.106 mmol), LiCl (89.9 mg, 2.12 mmol), diethylamine (2.19 ml, 21.2 mmol), and Pd(PPh₃)₄ (62.0 mg, 0.053 mmol). Then a mixture of the carbonate (365.0 mg, 1.06 mmol) and tetrahydropyranyl ether of propargyl alcohol (148.4 mg, 1.06 mmol) in THF (2 ml) was added. After being stirred at a room temperature for 30 min, the reaction mixture was diluted with hexane (30 ml). After the usual work-up, the allenyl acetylene was isolated by column chromatography as a pale yellow oil (306 mg, 71%) as an inseparable mixture of four diastereomers. The structure was fully supported by its ¹H NMR(400 MHz) and ¹³C NMR(100 MHz). Anal. Calcd. for C₂₃H₃₄O₆: C, 67.96; H, 8.43. Found: C, 67.75; H, 8.51.

Results of the reactions of several 2-alkynyl carbonates **1** with terminal acetylenes are shown in the table. The smooth reaction was observed with substituted 2-alkynyl carbonates. Although the reactants disappeared rapidly, no stable coupling product could be isolated when either R¹ or R² is hydrogen.



Entry	R ¹	R ²	R ³	R ⁴	Yield (%) ^a
1	-(CH ₂) ₅ -		n-Bu	CH ₂ OH	31
2		-(CH ₂) ₃ -	n-Bu	CH ₂ OSiMe ₂ ^t Bu	60
3	n-C ₆ H ₁₃	CH ₃	H	CH ₂ OTHP	65 ^b
4	n-C ₆ H ₁₃	CH ₃	n-Bu	CH ₂ OTHP	69
5	n-C ₆ H ₁₃	CH ₃	CH ₂ OTHP	CH ₂ OTHP	60
6	n-C ₆ H ₁₃	CH ₃	CH ₂ OTHP	n-Bu	42
7		CH ₃	n-Bu	CH ₂ OTHP	83
8		CH ₃	n-Bu	CH ₂ OH	62
9		CH ₃	CH ₂ OTHP	CH ₂ OTHP	71
10		CH ₃	CH ₂ OTHP	CH ₂ OH	66

a) Carbonate **1** (0.5 mmol), acetylene (0.5 mmol), CuI (0.05 mmol), Et₂NH (10.0 mmol), LiCl (1.0 mmol), Pd(PPh₃)₄ (5 mol%) were used.

The reaction was carried out at room temperature for 0.5 h.

b) Carbonate (1.0 mmol), acetylene (2.0 mmol), CuI (1.0 mmol), Et₃N (1.0 mmol), LiCl (10.0 mmol), Pd(OAc)₂ (5 mol%), and tri-*o*-tolylphosphine (20 mol%) were used. The reaction was carried out at 0°C for 1h.

References.

- 1) a) J. Tsuji, H. Watanabe, I. Minami, I. Shimizu, *J. Am. Chem. Soc.*, **107**, 2196(1985). b) I. Minami, Y. Yuhara, H. Watanabe, J. Tsuji, *J. Organometal. Chem.*, **334**, 225(1987). c) J. Tsuji, T. Sugiura, I. Minami, *Tetrahedron Lett.*, **27**, 731(1986).
- 2) a) W. D. Celmer, I. A. Solomons, *J. Am. Chem. Soc.*, **75**, 1372(1953), b) R. E. Bew, J. R. Chapman, E. R. H. Jones, B. E. Lowe, G. Lowe, *J. Chem. Soc. C*, 1966, 129 and 135.
- 3) For example, R. Nagata, H. Yamanaka, E. Murahashi, I. Saito, *Tetrahedron Lett.*, **31**, 2907(1990), and references cited therein.
- 4) a) T. Jeffery-Luong, G. Linstrumelle, *Synthesis*, 1983, 32. b) K. Ruitenbergh, H. Kleijn, C. J. Elsevier, J. Meijer, *Tetrahedron Lett.*, **22**, 1451(1981).
- 5) D. Sonogashira, Y. Tohda, N. Hagihara, *Tetrahedron Lett.*, 1975, 4467.

(Received in Japan 19 September 1990)